

Polyatomic Ions					
ion	formula	ion	formula	ion	formula
americyl	AmO_2^{2+}	acetate	CH_3COO^-	tetraborate	$\text{B}_4\text{O}_7^{2-}$
carbonyl	CO^{2+}	amide	NH_2^-	carbide	C_2^{2-}
thiocarbonyl	CS^{2+}	hydroxylamide	NHOH^-	carbonate	CO_3^{2-}
chromyl	CrO_2^{2+}	azide	N_3^-	chromate	CrO_4^{2-}
neptunyl	NpO_2^{2+}	hydrazide	N_2H_3^-	dichromate	$\text{Cr}_2\text{O}_7^{2-}$
plutonyl	PuO_2^{2+}	bromate	BrO_3^-	imide	NH^{2-}
seleninyl	SeO^{2+}	chlorate	ClO_3^-	molybdate	MoO_4^{2-}
selenonyl	SeO_2^{2+}	cyanide	CN^-	peroxide	O_2^{2-}
thionyl/sulfinyl	SO^{2+}	cyanate	OCN^-	oxalate	$\text{C}_2\text{O}_4^{2-}$
sulfonyl/sulfuryl	SO_2^{2+}	thiocyanate	SCN^-	phthalate	$\text{C}_8\text{H}_4\text{O}_4^{2-}$
uranyl	UO^{2+}	selenocyanate	SeCN^-	selenate	SeO_4^{2-}
vanadyl	VO^{2+}	tellurocyanate	TeCN^-	disulfide	S_2^{2-}
ammonium	NH_4^+	hydroxide	OH^-	sulfate	SO_4^{2-}
hydronium	H_3O^+	iodate	IO_3^-	thiosulfate	$\text{S}_2\text{O}_3^{2-}$
iodyl	IO_2^+	methanolate	CH_3O^-	dithionate	$\text{S}_2\text{O}_4^{2-}$
nitrosyl	NO^+	methanethiolate	CH_3S^-	silicate	SiO_3^{2-}
thionitrosyl	NS^+	ethanolate	$\text{C}_2\text{H}_5\text{O}^-$	borate	BO_3^{3-}
phosphoryl	PO^+	permanganate	MnO_4^-	arsenate	AsO_4^{3-}
thiophosphoryl	PS^+	nitrate	NO_3^-	phosphate	PO_4^{3-}
phospho	PO_2^+	superoxide	O_2^-	orthosilicate	SiO_4^{4-}